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glass substrate using a variety of process methods and materials. One of the more widely used of the latter processes is the electroless (autocatalytic) plating process. This process utilizes a basic electroless plating solution comprising a tin salt, in combination with a catalyst composition comprising a palladium salt, in combination with a reducing agent such as formaldehyde. The plating solution is deposited in the openings of a patterned metallic foil (also referred to as a "seed", "catalyst", or "plating matrix") which has previously been applied to the glass substrate by a variety of processes. In the plating process, reduction of the palladium (IV) ions in the catalyst solution leads to the catalytic activity of the salt. The plating solution catalyzes electroless deposition of palladium on the glass substrate as well as in the openings of the seed foil. The plating solution is also used to catalyze electroless deposition of copper on the palladium-coated glass substrate. The substrate is then immersed in an electroplating solution to

deposit copper on the substrate and in the openings of the seed foil. The copper on the substrate is in the form of a continuous conductive path, and the silver on the seed layer is in the form of a discontinuous conductive path. The discontinuities between the copper and silver paths form the plated interconnects. A wide variety of glass compositions has been used as the substrate material for the interconnect and/or pl 1cb139a0ed

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